Reg. No.

Question Paper Code : 11270

B.E./B.Tech. DEGREE EXAMINATION, MAY/JUNE 2016

Second Semester

Civil Engineering

HS 1153 – ENGINEERING CHEMISTRY – II

(Common to all branches)

(Regulations 2008)

Time : Three Hours

Maximum : 100 Marks

Answer ALL questions. PART – A $(10 \times 2 = 20 \text{ Marks})$

- 1. Define refractoriness under load ? What is its use ?
- 2. What are solid lubricants ? Give two examples.
- 3. Humidity of air is a culprit for corrosion, substantiate.
- 4. What is electroless plating ?
- 5. What is vulcanization of rubber?
- 6. List out any four differences between addition and condensation polymerization.
- 7. How is water gas superior to producer gas ?
- 8. Define the terms cracking and knocking.
- 9. Comprehend SMILEYS notation.
- 10. What is the significance of bond characteristics in drug design ?

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PART - B (5 × 16 = 80 Marks)

11.	(a)	(i)	What are the characteristics of good lubricants ? Explain the lubrication mechanism.	(8)
		(ii)	Describe the manufacture and characteristics of magnesite bricks. OR	(8)
	(b)	(i)	Write a note on silicon carbide and boron carbide.	(8)
	(0)	(ii)	Write briefly on the preparation, properties and uses of zirconia brick.	(8)
12.	(a)		at is corrosion ? What are its types ? Briefly explain the mechanism in each with suitable examples.	(16)
			OR	
¢	(b)	How cont	v is corrosion prevented ? Explain the important methods of corrosion rol.	(16)
13.	(a)	(i)	Discuss the preparation, properties and uses of Teflon and Nylon 6:6.	(8)
		(ii)	Explain the process of vulcanization of rubber. What are its advantages ?	(8)
	1 P	5	OR OR	
	(b)	(i)	Discuss the preparation, properties and uses of Bakelite.	(8)
2		(ii)	Explain injection moulding with a neat diagram.	(8)
14.	(a)	-	lain the petroleum refinery process in detail with neat sketches. What are the perties and applications of its various fractions ?	(16)
		2	OR	
	(b)	(i)	Describe the Fischer Tropsch method with a neat sketch.	(8)
		(ii)	What is producer gas? What is its composition? Discuss its method of production.	(8)
1.5		a		
15.	(a)	(i)	Elaborate on the relationship between structural information and the chemical property of molecules.	(8)
		(ii)	Define similarity search and sub-structure search.	(8)
			OR AND	1940
	(b)	(i)	Explain the terms, Canonical structure and structural keys.	(8)
		(ii)	Discuss the application of cheminformatics in drug designing.	(8)

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